

**Paper 1**

Questions are applicable for both core and extended candidates

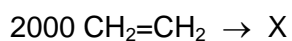
1 Which row explains why plastics such as poly(ethene) cause pollution?

	produce toxic gases when burned	accumulate in the oceans
<b>A</b>	no	no
<b>B</b>	no	yes
<b>C</b>	yes	no
<b>D</b>	yes	yes

2 Which row describes the relative sizes of monomer and polymer molecules?

	monomer	polymer
<b>A</b>	large	large
<b>B</b>	large	small
<b>C</b>	small	large
<b>D</b>	small	small

3 In reaction R, 2000 molecules of  $\text{CH}_2=\text{CH}_2$  react to form a single molecule X only.



Which terms describe reaction R,  $\text{CH}_2=\text{CH}_2$  and X?

	reaction R	$\text{CH}_2=\text{CH}_2$	X
<b>A</b>	addition	monomer	polymer
<b>B</b>	addition	polymer	monomer
<b>C</b>	substitution	monomer	polymer
<b>D</b>	substitution	polymer	monomer

4 Polymers are long-chain molecules made from small molecules linked together.

Four polymers or types of polymer are listed.

- 1 carbohydrates
- 2 nylon
- 3 proteins
- 4 *Terylene*

Which polymers or types of polymer are synthetic?

- A** 1 and 3      **B** 1 and 4      **C** 2 and 3      **D** 2 and 4

5 Many molecules of J join together in reaction R to form a long chain molecule K.

K is the only product.

Which row describes molecule J, reaction R and molecule K?

	molecule J	reaction R	molecule K
<b>A</b>	polymer	addition	monomer
<b>B</b>	monomer	addition	polymer
<b>C</b>	polymer	cracking	monomer
<b>D</b>	monomer	cracking	polymer

6 Some information about four substances, P, Q, R and S, is listed.

P is made by combining many small molecules together.

Molecules of Q are the largest molecules found in petroleum.

R is produced by cracking alkanes.

S is nylon.

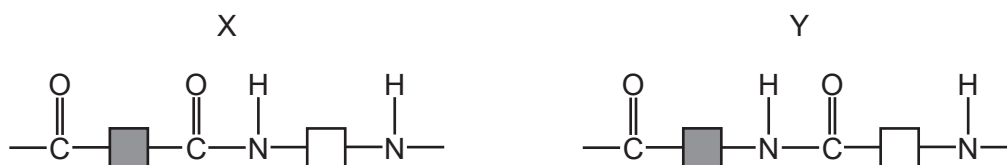
Which substances are synthetic polymers?

**A** P and Q      **B** P and S      **C** Q and R      **D** R and S

## Paper 2

**Questions are applicable for both core and extended candidates  
unless indicated in the question**

7 Parts of the structure of two different polymers, X and Y, are shown.

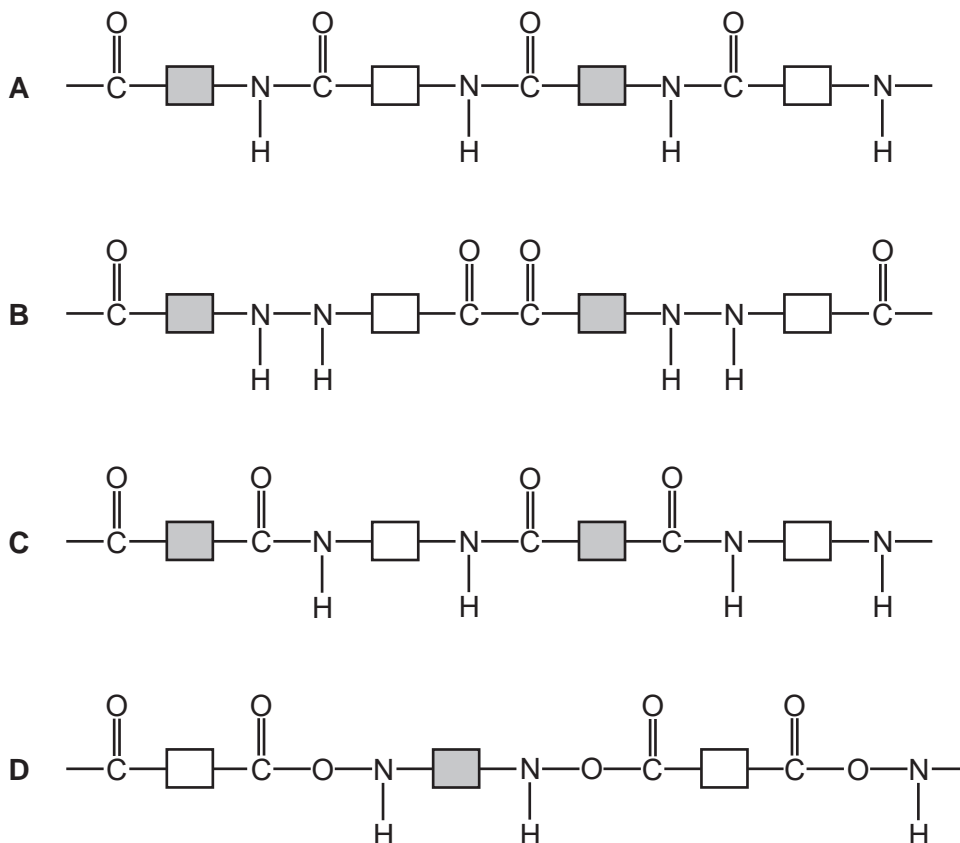


Which row about the monomers and the linkages between the monomers in polymers X and Y is correct? **(extended only)**

	monomers in X and Y	linkages
<b>A</b>	different	the linkages in X are different from the linkages in Y
<b>B</b>	different	the linkages in X are the same as the linkages in Y
<b>C</b>	same	the linkages in X are different from the linkages in Y
<b>D</b>	same	the linkages in X are the same as the linkages in Y

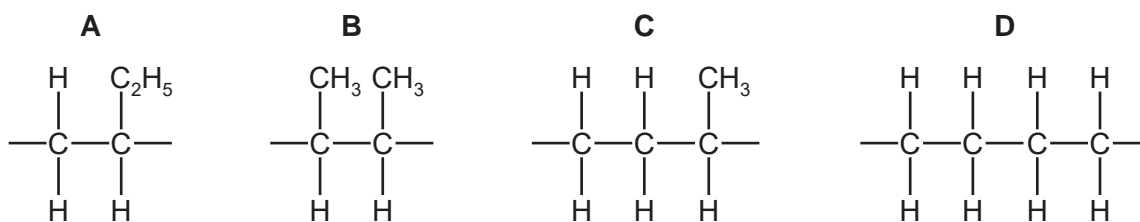
8 Nylon is formed by condensation polymerisation.

Which structure represents nylon? **(extended only)**

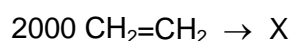


9 Which structure represents the repeat unit of the addition polymer formed from but-1-ene?

(extended only)



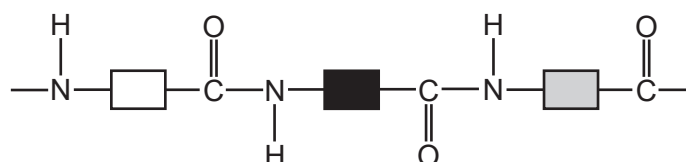
10 In reaction R, 2000 molecules of  $\text{CH}_2=\text{CH}_2$  react to form a single molecule X only.



Which terms describe reaction R,  $\text{CH}_2=\text{CH}_2$  and X?

	reaction R	$\text{CH}_2=\text{CH}_2$	X
<b>A</b>	addition	monomer	polymer
<b>B</b>	addition	polymer	monomer
<b>C</b>	substitution	monomer	polymer
<b>D</b>	substitution	polymer	monomer

11 Part of the structure of a polymer is shown.

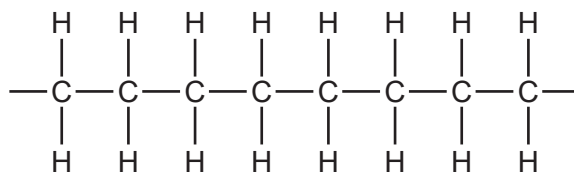


Which statements about the polymer are correct? (extended only)

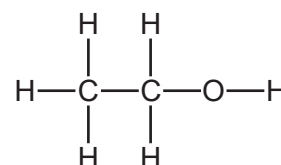
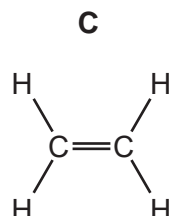
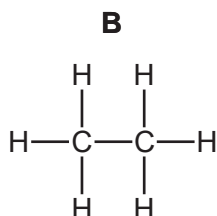
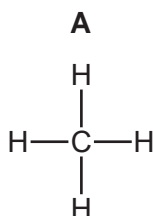
- The polymer is nylon.
- The polymer is formed by condensation polymerisation.
- There are ester linkages between the monomers.

**A** 1 and 2      **B** 2 and 3      **C** 2 only      **D** 3 only

12 The diagram shows part of a polymer.



Which diagram shows the monomer from which this polymer is made? (extended only)



13 Nylon and PET are polymers.

Which statements about these polymers are correct? (extended only)

- 1 They are both condensation polymers.
- 2  $\text{HOCH}_2\text{CH}_2\text{CH}_2\text{OH}$  could be a monomer for both polymers.
- 3 The complete combustion of both polymers gives two products only.

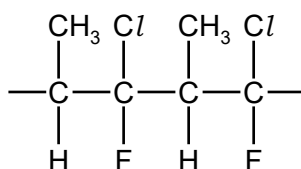
**A** 1 and 2

**B** 1 and 3

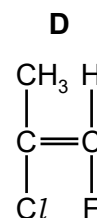
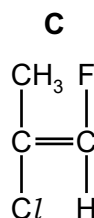
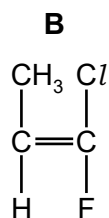
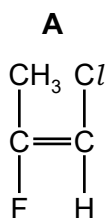
**C** 1 only

**D** 2 and 3

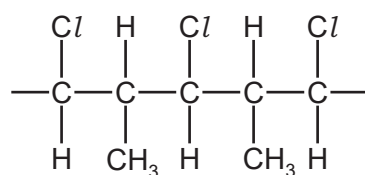
14 Part of the structure of a polymer is shown.



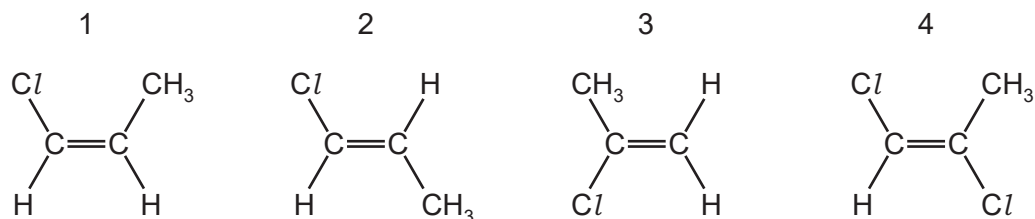
Which monomer is used to make this polymer? (extended only)



15 The structure of part of a polymer is shown.

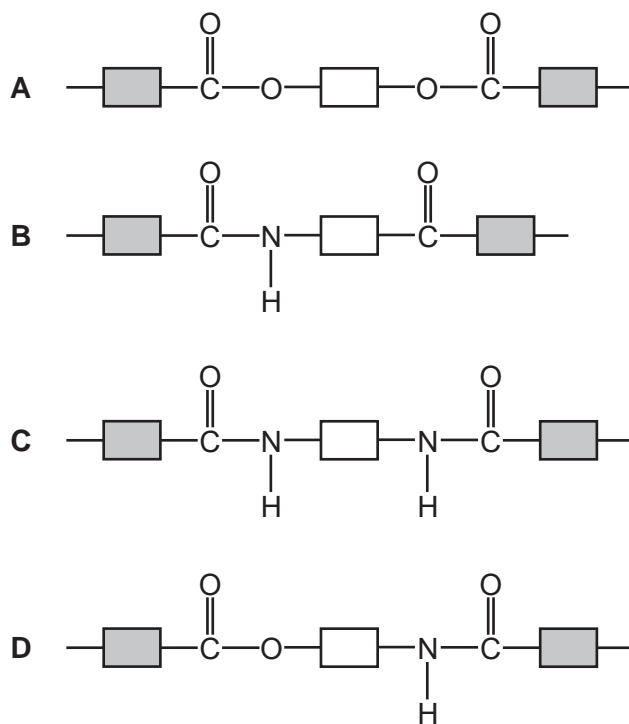


Which monomers can be used to make this polymer? (extended only)



- A** 1 and 2      **B** 1 and 4      **C** 2 and 3      **D** 3 and 4

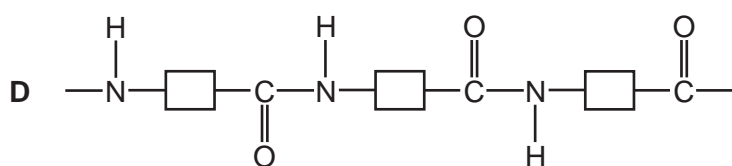
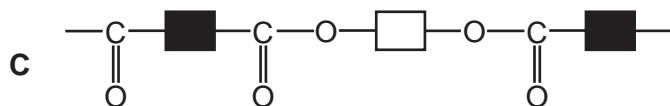
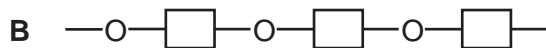
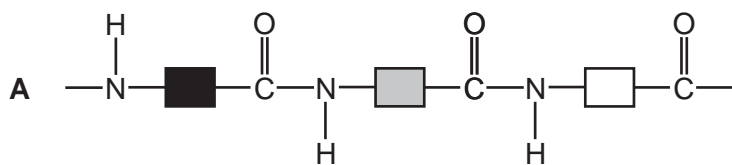
16 Which diagram represents the structure of nylon? (extended only)



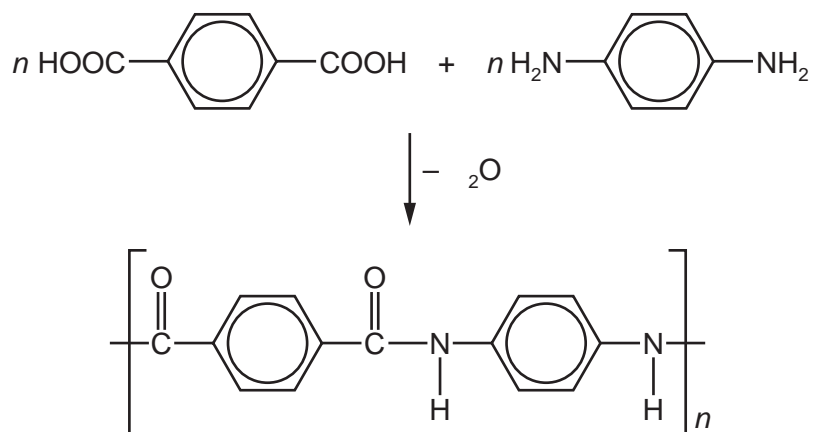
17 Which polymer is a synthetic polyamide? (extended only)

- A** nylon  
**B** poly(ethene)  
**C** protein  
**D** Terylene

18 Which structure represents *Terylene*? (extended only)



19 The equation shows the formation of a polymer called *Kevlar*.



Which row describes *Kevlar*? (extended only)

	how the polymer is formed	type of polymer
<b>A</b>	addition polymerisation	polyamide
<b>B</b>	addition polymerisation	polyester
<b>C</b>	condensation polymerisation	polyamide
<b>D</b>	condensation polymerisation	polyester